## Erratum

## Some Remarks on the Pariser-Parr-Pople Method

## Кіміо Онло\*

Theoret. chim. Acta (Berl.) 2, 219-227 (1964)

Received June 2, 1969/November 28, 1969

Wrong formulae were used for the two exchange integrals  $K(1s, 2p \sigma)$  and  $K(1s, 2p \pi)$  in the claculation whose results were reported in Tables 1, 2 and 3. The corrected results are listed in the following tables.

Table 1. Valence state energy of C (in eV)

	C <sup>+</sup>	С	C <sup>-</sup>
Electron configuration	$(sp^2)^3$	$(sp^2)^3\pi$	$(sp^2)^3\pi^2$
Orbital exponent of 1s	5.66	5.67	5.67
Orbital exponent of 2s	1.75	1.65	1.60
Orbital exponent of 2pσ	1.75	1.57	1.48
Orbital exponent of $2p\pi$	_	1.53	1.15
Total energy	-1007.98	-1016.67	-1012.38
σ-core energy	-1007.98	1006.95	-1005.81
W	_	- 9.72	- 9.43
$(\pi\pi \pi\pi)$	_	16.34	12.30

Table 2. Valence state energy of pyridine-type N (in eV)

	N <sup>+</sup>	N	N-
Electron configuration	$(sp^2)^4$	$(sp^2)^4\pi$	$(sp^2)^4\pi^2$
Orbital exponent of 1s	6.65	6.66	6.66
Orbital exponent of 2s	2.04	1.94	1.88
Orbital exponent of $2p\sigma$	2.07	1.87	1.71
Orbital exponent of $2p\pi$	_	1.90	1.60
Total energy	-1454.68	- 1465.55	- 1460.38
$\sigma$ -core energy	- 1454.68	-1453.16	- 1449.77
W	_	- 12.39	- 13.85
$(\pi\pi \mid \pi\pi)$	-	20.22	17.08

<sup>\*</sup> Present address: Department of Chemistry, Faculty of Science, Hokkaido University, Sapporo, Japan.

	N <sup>++</sup>	N <sup>+</sup>	N
Electron configuration	$(sp^2)^3$	$(sp^2)^3\pi$	$(sp^2)^3\pi^2$
Orbital exponent of 1s	6.65	6.65	6.66
Orbital exponent of 2s	2.17	2.06	1.97
Orbital exponent of $2p\sigma$	2.27	2.08	1.92
Orbital exponent of $2p\pi$	_	2.08	1.83
Total energy	-1428.05	-1454.48	-1462.10
$\sigma$ -core energy	-1428.05	-1426.97	-1424.26
W	_	- 27.51	- 28.64
$(\pi\pi \pi\pi)$	_	22.11	19.44

Table 3. Valence state energy of pyrrole-type N (in eV)

The usual assumption in the Pariser-Parr-Pople method

$$W(n, 1) = W(n, 2) = -I_n$$

is valid within an error of about 1.7 eV. Thus, the general conclusions reached in the original article remain essentially valid.

The author wishes to thank Dr. H. J. Silverstone and Dr. H. W. Joy whose remarks in their papers brought the error to his attention. He is also grateful to Mr. H. Narumi for his help in numerical calculations.

Professor Kimio Ohno Department of Chemistry Faculty of Science Hokkaido University, Sapparo, Japan